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(54) Title: UNSATURATED ETHYLENE POLYMERS

## (57) Abstract

Ethylene alpha, internal nonconjugated diene unsaturated copolymers and higher interpolymers having structural characteristics that provide improved physical properties and processability. The ethylene polymers of the invention have narrow molecular weight distribution and/or narrow composition distribution and/or a low cluster index. Ethylene/1,4-hexadiene and ethylene/vinyl cyclohexene polymers have residual unsaturation for subsequent modifications and for applications with improved toughness, printability, melt processability, and other advantageous characteristics.

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UNSATURATED ETHYLENE POLYMERS1      BACKGROUND OF THE INVENTION

2            This invention relates to unsaturated ethylene polymers in  
3         general and in particular to such polymers having residual  
4         unsaturation, and formed by the polymerization of ethylene and an  
5         alpha, internal nonconjugated diene (diolefin) comonomer.  
6         Optionally, other monomers are also incorporated in the polyethylene  
7         structure.

8            Polymers consisting primarily of ethylene monomer are widely  
9         recognized materials for a great variety of uses in many fields.  
10        Certain other polymers have disadvantages for some applications  
11        because of the physical properties associated with the polymer  
12        structure, whether from the incorporation of a comonomer or from the  
13        method of polymerization of the polymer. Most homopolymers and  
14        copolymers of alpha-olefins are widely used because of their  
15        inexpensive nature and excellent physical properties including  
16        transparency, moldability, chemical resistance, and other aspects.  
17        However, such polymers generally do not have widespread suitability  
18        for other substrates, surfaces, and materials due to their non-polar  
19        structure. Accordingly, most such homopolymers, copolymers, and  
20        higher interpolymers are not generally suitable for coatings,  
21        printing, and antistatic applications. Various methods and  
22        treatments have been devised to overcome the general shortcoming of  
23        such polymers. These include treating the polymers with oxidizing  
24        agents, degrading agents, and the like, but such measures have not  
25        always proven satisfactory for all applications.

26           The present invention is related to ethylene homopolymers,  
27        copolymers, terpolymers, and higher interpolymers of ethylene and  
28        other copolymerizable comonomers which have alpha, internal  
29        nonconjugated unsaturation. Such polymers always include a comonomer  
30        moiety which provides unsaturation in the polymer, the unsaturation  
31        being for subsequent modification or use in the various applications  
32        for which the ethylene polymers are not otherwise suitable. Polymers  
33        of ethylene are needed which not only incorporate residual  
34        unsaturation but also have outstanding physical properties above  
35        those commonly found in the predecessors of the present invention.

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1     Ethylene and other alpha-olefins are traditionally polymerized or  
2     copolymerized in the presence of catalytic complexes formed from  
3     mixtures of transition metal compounds and aluminum alkyls. Such  
4     complexes necessarily produce polymers and interpolymers which have  
5     physical properties depending on the catalyst and method of  
6     polymerization.

7       Prior art polymers have suffered from one or more  
8     difficulties affecting their physical properties, especially for the  
9     residual unsaturation polymers to which the present invention is  
10    related since these polymers may not often be modified or degraded  
11    without affecting the residual unsaturation content. The  
12    difficulties include broadened molecular weight distributions,  
13    broadened composition distribution of comonomer, and inefficient  
14    distribution or dispersion of the comonomer along the polyethylene  
15    chain of the copolymer. Of course, it is recognized that the  
16    polymers of the invention have residual unsaturation along the side  
17    chains substituted on the main polyethylene chain of the invention.

18       Broadened molecular weight distribution of a polymer  
19    strongly influences its melt flow properties and such polymers tend  
20    to have a high concentration of high molecular weight molecules  
21    making them subject to orientation. As a result, such resins produce  
22    strongly anisotropic physical properties in the machine transverse  
23    direction of a fabrication process and such properties are  
24    detrimental to a number of end use applications.

25       Broad molecular weight distribution resins also frequently  
26    contain a significant portion of quite low molecular weight  
27    material. These molecules almost invariably contain high  
28    concentrations of comonomer and therefore tend to be amorphous or of  
29    a low degree of crystallinity. Consequently, these materials exude  
30    to the surface of fabricated parts causing tackiness where not  
31    desired and/or interfere with other additives in the polymer designed  
32    according to the particular application. An example of this is the  
33    surface active property associated with slip agents in blown or cast  
34    film.

35       Most prior art copolymers tend to have a very broad  
36    composition distribution of comonomer, i.e., the distribution of  
37    comonomer among the polymer molecules being nonuniform, some

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1       molecules having a relatively high concentration of comonomer while  
2       others have a relatively low concentration of comonomer. This  
3       structural property of the prior art polymers allows that portion  
4       with low comonomer content to have a high melting point and vice  
5       versa leading to a broad melting range for the entire polymer  
6       composition. Of course the presence of a high melting component is  
7       disadvantageous for many applications where softness is desired and  
8       may lead to undesired stiffness. On the other hand, the presence of  
9       high comonomer content materials of low melting point frequently  
10      results in a high quantity of extractables.

11      Prior art materials are generally characterized by  
12      relatively inefficient use of comonomer moieties along the  
13      polyethylene chains. The distribution of the comonomer along the  
14      chain is very important in determining the efficiency of use of the  
15      comonomer and the resulting properties of the polymer, especially  
16      with respect to crystallizability of the polymer. Such prior art  
17      polymers tend to have a high degree of clustering of the comonomer  
18      moieties along the polyethylene chain. That is, the comonomer units  
19      are adjacent or nonisolated along the chain resulting in inefficient  
20      use of the comonomer since only a single polyethylene chain  
21      interruption results when several comonomer units are contiguous.  
22      This has extremely important implications in the total amount of  
23      comonomer required to achieve the desired crystallinity. Also, it is  
24      often disadvantageous to include unneeded portions of comonomer,  
25      especially when dealing with less available and more expensive diene  
26      comonomers. Frequently, the requirement to use a higher portion of  
27      comonomer also tends to force the comonomer into low molecular  
28      weight, high comonomer content ends of the molecular weight  
29      distribution. Accordingly, improved unsaturated ethylene copolymers,  
30      terpolymers, and interpolymers in the entire range of amorphous to  
31      highly crystalline polymers is needed.

32      BRIEF DESCRIPTION OF THE DRAWINGS

33      Figure 1 is a graphic and verbal depiction of the concept of  
34      cluster index;

35      Figure 2 is a plot of cluster index against mole percent  
36      comonomer for several commercial resins and the unsaturated resin of  
37      the invention;

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1       Figure 3 is an overlay of a plot of the molecular weight  
2       distribution of resin of the invention over that for commercial resin  
3       3, plotted against weight percent comonomer content;

4       Figure 4 is a series of plots of comonomer distribution in weight  
5       percent versus elution temperature for resin of the invention and  
6       several commercial resins;

7       Figure 5 is a plot evidencing the correlation of elution  
8       temperature and polymer composition in branches/1000 carbon atoms  
9       (mole percent comonomer);

10      Figure 6 is a plot for comparison to Figure 6A, showing the  
11      relative breadth of melting point for some commercial polymers;

12      Figure 6A is a plot of the melting point by DSC for three  
13      copolymers of the invention (Examples 1- 2A).

14      SUMMARY OF THE INVENTION

15      The present invention is directed to providing polymers of  
6       ethylene and at least one comonomer wherein the at least one  
17       comonomer is an alpha, internal nonconjugated diolefin and wherein  
18       the polymer has excellent physical properties resulting from the  
19       structure of the polymer and also has residual unsaturation. The  
20       polymers of the invention generally comprise at least about 50  
21       percent ethylene and the remainder alpha, internal nonconjugated  
22       diolefin and optionally other polymerizable comonomer content. When  
23       used herein, the term copolymer is intended to include terpolymers  
24       and higher interpolymers of ethylene as well as the simple copolymer  
25       of ethylene with a single diolefin or mixture of such alpha, internal  
26       nonconjugated diolefins. Therefore, the term copolymer includes  
27       terpolymers and interpolymers when referring to the polymers of the  
28       present invention.

29      The polymers of the present invention may have a broad range  
30       of densities and, therefore, applications for the particular type of  
31       copolymer. In addition, the residual unsaturation of the copolymer  
32       may vary quite widely by design according to the invention so as to  
33       provide a copolymer of particular suitability for certain  
34       applications. The polymers may be liquid or solid.

35      DESCRIPTION OF THE PREFERRED EMBODIMENTS

36      A copolymer composition from the polymerization of ethylene  
37       and at least one polymerizable comonomer comprising an alpha,

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1 internal nonconjugated diene, said copolymer having a composition  
2 distribution wherein at least about 55 weight percent, preferably  
3 about 65 weight percent, more preferably about 70 weight percent of  
4 the copolymer molecules have a comonomer content within about 50  
5 percent of the median comonomer content in mole percent of said  
6 copolymer composition.

7 By way of explanation, a copolymer having a median (equal  
8 amount greater and less than) comonomer content of 2.0 mole percent  
9 would have 55 weight percent, preferably 65, more preferably 70  
10 weight percent of its composition with a comonomer content of 1.0 to  
11 3.0 mole percent.

12 Another preferred embodiment of the present invention is a  
13 copolymer composition from the polymerization of ethylene and at  
14 least one polymerizable comonomer comprising an alpha, internal  
15 nonconjugated diene, said copolymer having a molecular weight  
16 distribution ( $M_w/M_n$ ) of about 3.0 or less, preferably 2.5 or less,  
17 preferably about 2.0.

18 Another preferred embodiment of the present invention is a  
19 copolymer composition from the polymerization of ethylene and at  
20 least one comonomer comprising an alpha, internal nonconjugated  
21 diene, said copolymer composition having incorporated in its  
22 structure at least 3 mole percent preferably 5 mole percent, more  
23 preferably 10 mole percent of units of said at least one comonomer,  
24 said copolymer composition having a cluster index of about 9 or less,  
25 preferably between 5 and 0.

26 The ethylene copolymers of the invention have improved  
27 properties resulting especially from the more efficient use of diene  
28 comonomer in controlling the crystallizability of the polymer. That  
29 is, the efficient use of the diene comonomer comprises an improved  
30 isolation of the comonomer molecules along the polyethylene chain as  
31 not previously achieved for ethylene copolymers having residual  
32 unsaturation. Accordingly, the polymers of the present invention not  
33 only have especially good application for those uses previously  
34 employing residual unsaturation polymers, but also have excellent  
35 overall physical properties marking a significant improvement over  
36 those materials previously available. The improved properties of the  
37 invention result from the isolated dispersion of the diene comonomer

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1 and other comonomers along the sequence of the polymer molecule of  
2 the invention.

3 Another preferred embodiment copolymer of the present  
4 invention has a distribution wherein more than 55 weight percent of  
5 the copolymer molecules are within 50 percent of the median comonomer  
6 content in the mole percent, said copolymer being formed by  
7 polymerization in the presence of a catalyst system comprising a  
8 metallocene of a metal of Group IVB, VB, and VIB of the Periodic  
9 Table and an alumoxane or reaction product thereof.

10 The unsaturated copolymer products of the present invention  
11 comprise batchwise or continuously produced bulk polymer compositions  
12 having the properties and characteristics described herein. No such  
13 compositions have heretofore been discovered. That is, the  
14 entire/unmodified polymerization product has the advantageous  
15 properties.

16 The unsaturated copolymer compositions of the invention may  
17 be prepared from the polymerization of ethylene and at least one  
18 comonomer. The at least one comonomer comprises, that is always  
19 includes some (or all) of the alpha, internal nonconjugated diene.

20 The alpha, internal nonconjugated diene comonomers of the  
21 invention may be linear or cyclic, branched or unbranched, short or  
22 long chained.

23 The diene comonomers incorporated in the unsaturated  
24 polymers of the invention may be in small or large quantity relative  
25 to the amount of ethylene in the polymer. In one embodiment of the  
26 invention, the polymers of the invention contain at least a minimum  
27 of about 3 mole percent total comonomer based on the moles of  
28 ethylene and comonomers, so as to provide wide dispersion of the  
29 comonomers in the polymer product composition. This generally limits  
30 the density to a number below 0.930 g/cc, preferably below 0.920 g/cc  
31 depending on the copolymers chosen and the method of incorporation.  
32 Thus, where only a two component polymer composition is formed from  
33 ethylene and the alpha, internal nonconjugated diene, at least about  
34 3 mole percent diene units and no more than about 97 mole percent  
35 ethylene units are present. For terpolymers and higher interpolymers  
36 of the invention, only about 0.01 mole percent or more, preferably  
37 about 0.1 mole percent or more, more preferably 1 mole percent or

1 more of the diene need be incorporated so long as the total  
2 incorporation of comonomers (diene and other[s]) is at least about 3  
3 mole percent for this embodiment.

4 Despite incorporation of at least about 3 mole percent  
5 comonomer units in one embodiment of the polymer compositions of the  
6 invention the polymers nevertheless have a low cluster index and  
7 preferably other characteristics described herein. The total  
8 comonomer content of the polymers of the invention may be the  
9 predominant portion of the polymer. Preferably, especially for solid  
10 unsaturated polymers, the ethylene units are the predominant  
11 component on a molar basis.

12 In one preferred embodiment of the present invention the  
13 polymers of the invention have improved properties attributable to  
14 the more uniform content of diene and other comonomer among the  
15 polymer molecules.

16 In contrast to the prior art polymers, the copolymers,  
17 terpolymers, and other interpolymers of the present invention have  
18 residual unsaturation and exhibit very little clustering of the  
19 comonomer molecules along the polyethylene chain both with respect to  
20 the diene comonomer and any other polymerizable comonomer formed in  
21 the polyethylene chain. As a result, the use of comonomer in forming  
22 the copolymers of the invention is very efficient in controlling  
23 crystallizability, wards against formation of high comonomer  
24 content/low molecular weight ends, reduces cost, and improves  
25 properties.

26 In a preferred embodiment of the present invention, the  
27 copolymers of the invention have very narrow composition distribution  
28 of comonomer. That is, the copolymers have much more uniform  
29 distribution of comonomer among the molecules thereby largely  
30 avoiding the problem presented by broad composition distribution  
31 resins.

32 In another preferred delineation of the invention, the  
33 copolymers of the invention have improved properties resulting from  
34 the more consistent size of polymer molecules not achieved in the  
35 prior art. This aspect of the more preferred embodiment is generally  
36 referred to as the improved molecular weight distribution or ratio of  
37 the weight average molecular weight to number average molecular

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1        weight of the polymer material.

2        Also in this preferred embodiment of the present invention,  
3        the copolymers of the invention exhibit relatively narrow molecular  
4        weight distribution. That is, they have a relatively low ratio of  
5        weight to number average molecular weight. In other words, the  
6        concentration of very high molecular weight molecules and very low  
7        molecular weight molecules is decreased over prior art unsaturated  
8        polymers. The absence of the high molecular weight molecules reduces  
9        the tendency for orientation in fabrication processes and increases  
10      isotropic machine/transverse direction physical properties. The  
11      absence of a low molecular weight molecules (low tail) in the  
12      copolymers of the invention reduces the tendency to form tacky  
13      surfaces and otherwise interfere with surface active agents in  
14      certain applications of the copolymers of the present invention.

15      Unsaturated polymers of the prior art do not have the  
16      structure and corresponding properties of the polymers of the  
17      invention. That is, they lack the low cluster index for ethylene  
18      copolymers (including terpolymers and higher interpolymers) having  
19      therein at least about 3 mole percent, preferably 5 mole percent,  
20      more preferably 10 mole percent comonomer units according to the  
21      invention. The absence of such structure is generally reflected in  
22      easily measured characteristics of the invention such as melting  
23      point temperature and the like. The prior art unsaturated polymers  
24      also generally lack narrow molecular weight distribution and narrow  
25      comonomer distribution.

26      The polymers of the present invention are capable of being  
27      fabricated into a wide variety of articles, as is known for  
28      homopolymers of ethylene and copolymers of ethylene and higher  
29      alpha-olefins. Furthermore, the polymers of the present invention  
30      typically have residual unsaturation which permits their use in  
31      additional applications where the residual unsaturation may be used  
32      for bondability, printability, crosslinkability, and the like.

33      The unsaturated polymers of the invention may vary in  
34      density across a broad scale from essentially amorphous materials to  
35      highly crystalline materials. They may be liquids (such as for  
36      certain rubbers and lubricants and waxes) or solids.

37      The molecular weight of the copolymers of the invention may

1 vary over a broad range. Preferably the polymers have a number  
2 average molecular weight of about 500 or higher, preferably 1000 or  
3 higher, more preferably about 10,000 or higher. Typically, materials  
4 used for elastomers applications are either copolymers or terpolymers  
5 (often with propylene monomer) in the density range of about  
6 0.86-0.87 g/cc. Typically, these polymers contain 30 weight percent  
7 or more of the comonomers and the balance ethylene. Frequently, the  
8 elastomers of the unsaturated polymers of the invention will have as  
9 much as 48 weight percent of the comonomer present.

10 The unsaturated polymers of the present invention may also  
11 include plastomers in the density range of about 0.87-0.900 g/cc and  
12 containing from about 20-30 weight percent of comonomer. Also  
13 available in the unsaturated polymers of the invention are the very  
14 low density polyethylene materials of density range 0.900-0.915 g/cc  
15 and having from about 10-20 weight percent comonomer present.

16 The unsaturated polymers of the invention may be formed as a  
17 linear low density polyethylene type polymer in the density range of  
18 about 0.915 g/cc to about 0.940 g/cc and containing from about 5-10  
19 weight percent of the comonomers. The unsaturated polymers of the  
20 invention may also be used in the form of high density polyethylene  
21 having a density of about 0.940 g/cc and above and containing up to  
22 about 5 weight percent comonomers. The unsaturated polymers of the  
23 invention may also form amorphous materials below a density of 0.86  
24 g/cc including tackifier resins.

25 The unsaturated polymers of the present invention have  
26 particularly advantageous properties because of their narrow  
27 molecular weight distribution, narrow compositional distribution, and  
28 their chain configuration having isolated comonomer units.

29 Sequence distributions, or the distribution of comonomer  
30 units along a polymer chain, in, e.g., linear low density  
31 polyethylenes is a factor affecting the cost of the polymer because  
32 it affects the amount of comonomer required to achieve a desired  
33 polymer density. If comonomer is efficiently incorporated, i.e.,  
34 with little clustering of comonomer units, in a linear low density  
35 polyethylene, less comonomer is required to depress the density.  
36 Thus, the number of comonomer runs in a copolymer chain and the  
37 length of each run is significant in the structure of the

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1 polyethylene molecules and affects the physical properties of the  
2 polymer. The polymers of the present invention are marked by a  
3 relatively high number of single units of comonomer in the polymer  
4 chain relative to the number of units containing more than a single  
5 comonomer molecule in comparison to unsaturated polymers of the prior  
6 art.

7 The unsaturated ethylene polymers of the invention are  
8 hereinafter described by their "cluster index". This index reflects  
9 the degree to which the polymers of the invention have individual  
10 comonomer units dispersed along the polyethylene chain, preferably in  
11 favoring isolation of individual units over groups of two or more  
12 units. Given a minimum level of comonomer, the unsaturated ethylene  
13 polymers of the present invention are especially noted for their  
14 efficient use of comonomer molecules by having more isolated  
15 comonomer molecules along the polyethylene chain and fewer clusters  
16 of molecules of the comonomer in the polyethylene chain. That is,  
17 the unsaturated polymers of the present invention tend to deviate  
18 from random comonomer distribution in the direction of fewer  
19 contiguous comonomer sequences. Thus, the cluster index permits a  
20 quantitative evaluation of the deviation from a random distribution  
21 of comonomer in the polymer chain.

22 In the cluster index description given herein, there are two  
23 reference points. The reference point 0 describes a polymer which  
24 has only isolated comonomer insertions without any contiguous  
25 comonomer units in a cluster; of course this describes pure  
26 homopolymers also. The second reference point is the number 10  
27 describing an ethylene copolymer having comonomer distribution that  
28 is exactly random (Bernoullian) and thus containing a predictable  
29 amount of contiguous comonomer units. Any polymer having a cluster  
30 index value greater than 10 contains proportionally more contiguous  
31 comonomer sequences than predicted by the random distribution. Any  
32 polymer having a cluster index value between 0 and 10 is indicated to  
33 have fewer contiguous sequences than a random distribution polymer  
34 (given a minimum of comonomer). These values are typically  
35 associated with the method of producing the polymer including the  
36 catalyst used and conditions of polymerization.

37 The cluster index comparisons are best made for polymers

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1 having comparable comonomer mole contents or densities. The  
2 measurement of the clustering of the comonomer along the polyethylene  
3 chain in a given polymer may be determined by a study using carbon 13  
4 nuclear magnetic resonance spectroscopy ( $C^{13}$  NMR). Using this tool  
5 for evaluation, the cluster index may be given as follows:

$$6 \quad \text{Cluster index} = 10 \left[ (\bar{x}) - (\bar{E}_X) \right] / [2(\bar{x})^2 - (\bar{x})^3]$$

7 where (X) is the mole % of total comonomer molecules in the  
 8 copolymer and (EXE) is the mole fraction of the triad segment of  
 9 3 monomer units containing ethylene-comonomer-ethylene. These  
 10 concentrations are easily measured using C<sup>13</sup> NMR.

The basis for the cluster index is further explained and exemplified below by the following discussion and in reference to drawing Figure 1.

Referring now to drawing Figure 1 the cluster index may be derived as follows, using the reference point 10 as random clustering expected in a polymerization and the reference point 0 as that point for no clustering in a polymer (no units of more than one comonomer molecule contiguous). Accordingly,

$$\text{Cluster index} = 10 - 10 \times \frac{[(\text{EXE})_{\text{observed}} - (\text{EXE})_{\text{random}}]}{[(X) - (\text{EXE})_{\text{random}}]}$$

where "X" is the mole % comonomer in the ethylene copolymer and EXE is the corresponding triad configuration of a single comonomer molecule contiguous to two ethylene molecules (units).

The term (EXE)<sub>random</sub> serves as a reference point and its value can be calculated from any suitable statistical model. In this case the Bernoullian model was chosen. For the Bernoullian model

$$28 \quad (\text{EXE})_{\text{random}} = [1 - X]^2[X].$$

29 Thus, substituting this value of  $(\text{EXE})_{\text{random}}$  in the formula for  
30 cluster index above gives:

1      Cluster index =  $10 [(\bar{X}) - (\text{EXE})_{\text{observed}}] / [2(\bar{X})^2 - (\bar{X})^3]$

2      Accordingly, it can be readily seen from drawing Figure 1 that those  
 3      polymers having more comonomer clustering than a random distribution  
 4      appear to the left of the reference point 10 and those having less  
 5      comonomer clustering than a random distribution appear between 0 and  
 6      10.

7      In Figure 2, polymers are plotted according to their cluster  
 8      index as described for Figure 1 and using the vertical axis to plot  
 9      mole percent of comonomer of the polymer samples. It is readily  
 10     apparent from drawing Figure 2 that those polymers of the invention  
 11     have a reduced cluster index (for a given density) over polymers of  
 12     the prior art such as those commercial polymers plotted and appearing  
 13     near or to the left of reference point 10 (random clustering). The  
 14     technique of using the  $C^{13}$  NMR to obtain the information for  
 15     cluster index is known to the skilled artisan.

16     Also shown for comparison in Figure 2 is the cluster index  
 17     of Dowlex 2088 octene LLDPE copolymer (Resin 9), Dowlex 2517 octene  
 18     LLDPE copolymer (Resin 10), and Union Carbide 7099 hexene LLDPE  
 19     copolymer (Resin 11).

20     The cluster index improvement in the polymers of the  
 21     invention versus polymers of the prior art is detectable at 3 mole  
 22     percent comonomer content, readily discernable at 5 mole percent, and  
 23     remarkable at about 10 mole percent or higher.

24     The above described cluster index may be considered to be of  
 25     the first order or based on the total number of comonomer units less  
 26     the isolated (EXE) comonomer units. A higher order cluster index can  
 27     also be measured and calculated as based primarily on the occurrence  
 28     of dimer segments (EXX) or (XXE) as observed. This measurement is  
 29     somewhat more discriminating at low comonomer mole percents (about  
 30     3). Thus, an EXX index may in like fashion be calculated as:

31

$$\text{EXX index} = \frac{[(\text{EXX})_{\text{observed}} - (\text{EXX})_{\text{Bernoullian}}]}{(\text{EXX})_{\text{Bernoullian}}}$$

32     EXX index =  $10 - 10 \times$

33

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1        Since  $(EXX)_{\text{Bernoullian}} = 2[E][X]^2$  and the (EXX) observed is  
2        based on both (EXX) and (XXE) units, readily measured by C<sup>13</sup>  
3        NMR, the EXX index is readily achieved. For such index a totally  
4        random polymer is measured as 10, a totally dimerless polymer is  
5        0 (no contiguous XX), and a polymer increasingly deficient in  
6        solitary units (EXE) will approach 20.

7        The EXX index is a second measure of polymer structure based  
8        directly on dimers and higher orders of contiguous X (EXX and XXE)  
9        observed; it may be used independently or together with the cluster  
10      index (EXE) to distinguish polymers.

11      The EXX index for commercial resins 3, 5, and 6 in the Table  
12      herein are 7.3, 12.4, and 15.0. Compare the terpolymer resin of  
13      Example 2 of the invention which at 3.9 mole percent comonomer has an  
14      EXX index of only 8.2.

15      The unsaturated ethylene polymers of the present invention  
16      are preferably marked by a relatively narrow molecular weight  
17      distribution in comparison to prior art polymers. The molecular  
18      weights and molecular weight distributions were determined using  
19      Waters 150C Gel Permeation Chromatographic Instruments. These  
20      instruments were equipped with refractive index detectors operated at  
21      145°C and at a solvent flow rate of 1 milliliter per minute. The  
22      solvent used was ultra-high purity grade 1,2,4-trichlorobenzene  
23      obtained from Burdick and Jackson Company. Prior to use, the solvent  
24      was filtered through a 0.5 micron filter and stabilized with 120 ppm  
25      BHT. Three Waters styragel columns were used with nominal porosities  
26      of 500, 10,000, and 1,000,000 angstroms. Each polymer sample was  
27      dissolved in trichlorobenzene solvent at 145°C to a concentration  
28      level of about 0.1 weight percent and thereafter filtered through a  
29      0.5 micron porous metal filter. About 300 microliters of this  
30      solution was then injected into the gel permeation chromatograph.  
31      Analysis time was typically 45 minutes. Calibration of the  
32      instrument for molecular weight determination was accomplished with  
33      the use of narrow molecular weight distribution polystyrene standards  
34      obtained from Toyo Soda Manufacturing Company. Sixteen of the  
35      standards were used ranging in molecular weight from 526 - 5.2 x  
36      10<sup>6</sup>. Molecular weight distributions of these standards were listed  
37      at 1.0-1.15 as measured by the ratio of weight average molecular

1 weight to number average molecular weight. These polystyrene  
2 molecular weight data were then converted to a polyethylene basis  
3 with the use of the Mark-Houwink equation and the use of the  
4 following constants:

5 for polyethylene  $K = 5.17 \times 10^{-4}$ ,  $a = 0.70$

6 for polystyrene  $K = 2.78 \times 10^{-4}$ ,  $a = 0.70$ .

7 Each sample was run twice and the calculated results average to yield  
8 the molecular weight information reported. The molecular weight  
9 distribution of the resin of Example 1 below was plotted for  
10 comparison on the same plot with that for Exxon LL3001 linear low  
11 density polyethylene resin mole percent comonomer on a scale of the  
12 log of molecular weight versus weight percent of the polymer in the  
13 molecular weight range. As can be seen from drawing Figure 3 the  
14 unsaturated polymer of the invention has a strictly narrower  
15 molecular weight distribution than the commercially available linear  
16 low density polyethylene.

17 The unsaturated polymers of the present invention preferably  
18 also have a narrow distribution of comonomer amongst molecules of the  
19 polymer in comparison to prior art polymers. For comparison, drawing  
20 Figure 4 reflects the narrow distribution of the unsaturated polymer  
21 of Example 1 in comparison to the relatively broad comonomer  
22 distribution polymers of the prior art. In Figure 4 the weight  
23 percent of copolymer (having a given comonomer content) is plotted  
24 against an elution temperature which directly reflects comonomer  
25 content. A better understanding of this preferred embodiment of the  
26 polymers of the present invention will be had by a review of Figure 4  
27 and the following discussion.

28 Crystalline copolymers may be fractionated by comonomer  
29 content over a temperature range from 0-120°C in tetrachloroethylene  
30 solvent. The compositions of the fractions of the copolymer are  
31 determined and a solution temperature versus composition calibration  
32 curve has been constructed based on the data obtained. Using this  
33 calibration curve, the temperature scale of the solubility  
34 distribution curve can be converted to a composition scale and a  
35 composition distribution curve is thus obtained as in drawing Figure  
36 4.

37 A machine has been assembled to automatically determine the

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1 solubility distribution curve of a crystalline copolymer. In the  
2 measuring instrument, a steel column is packed with small mesh glass  
3 beads and immersed in an oil bath whose temperature can be programmed  
4 over a temperature range from about 0°C-150°C. The  
5 tetrachloroethylene solvent may be prevented from boiling by  
6 operating the instrument at about 3 atmospheres pressure under an  
7 automatic regulator. A weighed amount of sample, usually about 1.6  
8 grams, is placed in a sample preparation chamber, sealed, and  
9 repeatedly evacuated and filled with argon. A metered volume of  
10 solvent is then pumped into the sample preparation chamber where it  
11 is stirred and heated to obtain a solution of about 1 percent  
12 concentration. A metered volume of this solution, usually about  
13 100 cc is then pumped into the packed column which has been  
14 thermostated at a high temperature of usually at least about 120°C.

15 The polymer solution sample is subsequently crystallized by  
16 cooling the polymer in the column to 0°C at a programmed rate of 5°C  
17 per hour. The column was then maintained at 0°C for at least an  
18 hour. Thereafter, the elution stage of the determination is started  
19 by pumping pure solvent through the column at a rate of 6 cc per  
20 minute. Effluent from the column passes through the reheater where  
21 it is heated to 120°C before passing through an IR detector used to  
22 measure the absorbance of the effluent stream. The infrared  
23 absorbents of the polymer carbon hydrogen stretching bands at about  
24 2960 centimeter<sup>-1</sup> serves as a continuous measure of the relative  
25 concentration of polymer in the effluent. After passing through the  
26 infrared detector the temperature of the effluent is reduced to about  
27 110°C and the pressure is reduced to 1 atmosphere before passing the  
28 stream into an automatic fraction collector. In the elution stage,  
29 the pure solvent is pumped through the column set at 0°C for one  
30 hour. This serves to flush polymer that has not crystallized during  
31 the crystallization stage out of the column so that the percent of  
32 uncrosslinked polymer can be determined from the infrared trace.  
33 The temperature is then programmed upward at 10°C per hour to 100°C  
34 and at 20°C per hour from 100°C to 120°C.

35 The compositions of fractions obtained from the various  
36 polymers were determined by infrared spectroscopy. The IR  
37 compositions are obtained from the intensity of the 1378(cm)<sup>-1</sup>

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1      methyl band, the thickness of the sample, and a calibration curve  
2      based on samples whose compositions were determined independently by  
3      C<sup>13</sup> NMR. No corrections for polymer end groups was made in  
4      obtaining compositions from infrared data.

5      Figure 5 is a plot of the elution temperature versus the  
6      comonomer content in mole percent comonomer (branches per 1000 carbon  
7      atoms) for the fractions of the sample. The curve "A" has been drawn  
8      through the points. Thus, curve A may be used as a correlation  
9      between elution temperature and composition of the polymer for  
10     temperatures greater than 0°C. The calibration curve is most  
11     accurate for fractions with number average molecular weights, M<sub>n</sub>  
12     ≥ 10,000 as determined by size exclusion chromatography.

13     As seen in drawing Figure 4, samples of polymers tend to  
14     have a peak at about 0°C elution temperature. This small peak  
15     represents the fraction of total polymer that is not crystallizable  
16     at the lowest temperature of the experiment (about 0°C).

17     In summary, the device and procedure described provide a  
18     plot of relative weight percent of polymer versus elution temperature  
19     which in turn may be correlated to composition and mole percent  
20     comonomer (branches per 1000 carbon atoms) in the polymer chain.  
21     Accordingly, drawing Figure 4 is an effective comparison of the  
22     comonomer distribution of the unsaturated polymers of the invention  
23     to that for certain commercial polymers. As can readily be seen, the  
24     composition distribution of the polymer of the invention example is  
25     quite narrow in comparison.

26     The alpha internal nonconjugated dienes which form units of  
27     the copolymers (including terpolymers and higher interpolymers) of  
28     the invention include a variety of such diolefins exemplified by but  
29     not limited to the following:

30        1,4-hexadiene; 1,4-heptadiene; 1,5-heptadiene;  
31        1,4-octadiene; 1,5-octadiene; 1,6-octadiene; 4-methyl-1,4-hexadiene,  
32        5-methyl-1,4-hexadiene, 4-ethyl-1,4-hexadiene,  
33        5-methyl-1,4-heptadiene; 5-ethyl-1,6-octadiene; 3-vinyl cyclohexene;  
34        4-vinyl cyclohexene; 3-vinyl cyclopentene; cis and trans  
35        1,5,9-decatriene; 5-vinyl norbornene; 1,4-dodecadiene.

36     Comonomers other than the dienes of the invention include  
37     polymerizable comonomers such as acids, esters, other olefins, and

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1 the like. Suitable olefins include the alpha olefins such as  
2 propylene, butene, hexene, octene, 4-methyl-pentene-1, and others.

3 According to the invention, the alpha-olefin content of the  
4 unsaturated ethylene polymers (copolymers) of the invention is less  
5 than 15 weight percent, preferably less than 10 weight percent.

6 In forming the polymers of the present invention it may be  
7 necessary to purify, or isolate the diene and other, optional  
8 comonomers of the invention in order to obtain incorporation thereof  
9 during polymerization. One such recommended technique is passing the  
10 comonomer, especially the diene, over alumina to remove extraneous  
11 materials.

12 Figures 6 and 6A show the melting point by differential  
13 scanning calorimetry results for the resins of the invention in  
14 comparison to some commercial resins. Melting point distributions  
15 were determined using a Perkin Elmer DSC-7 operated in the following  
16 fashion: About 5-6 mg of sample was heated to 160°C and held at that  
17 temperature for 5 minutes. The sample was then cooled at 10°C per  
18 minute to a temperature of 0°C and reheated at 10°C per minute. The  
19 melting distributions reported here were collected during this  
20 reheating at 10°C per minute.

21 Concentrations of unsaturation in ethylene based polymers  
22 was determined by comparisons of IR spectra of brominated versus  
23 original polymer specimens utilizing the following bands:

24 vinylene - 965 cm<sup>-1</sup>

25 vinyl - 909 cm<sup>-1</sup>

26 vinylidene - 888 cm<sup>-1</sup>

27 A better understanding of the invention will be had by a  
28 review of the above discussion in conjunction with the drawing  
29 Figures and the Examples given below. The best mode of the invention  
30 presently known is described herein.

31 The unsaturated polymers of the present invention are those  
32 having the narrow cluster index and preferably having the additional  
33 described structure of molecular weight distribution and comonomer  
34 distribution as well as properties herein described. Such polymers  
35 may be prepared by any of the known techniques of polymerization  
36 including solution, high pressure, and gas-phase polymerization  
37 processes.

38 The polymers of the invention may be formed by use of  
39 catalyst systems of the metallocene type. That is,

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1 cyclopentadienylide catalyst systems using a metallocene complex in  
2 conjunction with an alumoxane cocatalyst or reaction product thereof  
3 are suitable for preparing polymers of the invention. The  
4 metallocene catalyst may be represented by the general formula  
5  $(Cp)_m MR^n R'^p$  wherein Cp is a substituted or unsubstituted  
6 cyclopentadienyl ring; M is a Group IVB, VB, or VIB transition metal;  
7 R and R' are independently selected halogen, hydrocarbyl group, or  
8 hydrocarboxyl groups having 1-20 carbon atoms; m = 1-3, n = 0-3, p =  
9 0-3; and the sum of m + n + p equals the oxidation state of M.  
10 Various forms of the catalyst system of the metallocene type may be  
11 used for polymerization to achieve polymers of the present invention  
12 including those of the homogeneous or the heterogeneous, supported  
13 catalyst type wherein the catalyst and alumoxane cocatalyst are  
14 together supported or reacted together onto an inert support for  
15 polymerization by gas-phase, high pressure, or solution  
16 polymerization.

17 The cyclopentadienyls of the catalyst may be unsubstituted  
18 or substituted with hydrogen or hydrocarbyl radicals. The  
19 hydrocarbyl radicals may include alkyl, alkenyl, aryl, alkylaryl,  
20 arylalkyl, and like radicals containing from about 1-20 carbon atoms  
21 or 2 carbon atoms joined together to form a  $C_4-C_5$  ring.

22 Exemplary hydrocarbyl radicals include methyl, ethyl, propyl, butyl,  
23 amyl, isoamyl, hexyl, isobutyl, heptyl, octyl, nonyl, decyl, cetyl,  
24 2-ethylhexyl, phenyl and the like. Exemplary halogen substituents  
25 include chlorine, bromine, fluorine and iodine and of these halogen  
26 atoms, chlorine is preferred. Exemplary hydrocarboxy radicals are  
27 methoxy, ethoxy, propoxy, butoxy, amyloxy and the like.

28 Illustrative, but non-limiting examples of the metallocene catalyst  
29 useful in preparing the polymers of the present invention include  
30 bis(cyclopentadienyl)titanium dimethyl, bis(cyclopentadienyl)titanium  
31 diphenyl, bis(cyclopentadienyl)zirconium dimethyl,  
32 bis(cyclopentadienyl)zirconium diphenyl, bis(cyclopentadienyl)hafnium  
33 dimethyl and diphenyl, bis(cyclopentadienyl)titanium di-neopentyl,  
34 bis(cyclopentadienyl)zirconium di-neopentyl,  
35 bis(cyclopentadienyl)titanium dibenzyl,  
36 bis(cyclopentadienyl)zirconium dibenzyl,  
37 bis(cyclopentadienyl)vanadium dimethyl; the mono alkyl metallocenes  
38 such as bis(cyclopentadienyl)titanium methyl chloride,

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1      bis(cyclopentadienyl)titanium ethyl chloride,  
2      bis(cyclopentadienyl)titanium phenyl chloride,  
3      bis(cyclopentadienyl)zirconium methyl chloride,  
4      bis(cyclopentadienyl)zirconium ethyl chloride,  
5      bis(cyclopentadienyl)zirconium phenyl chloride,  
6      bis(cyclopentadienyl)titanium methyl bromide,  
7      bis(cyclopentadienyl)methyl iodide, bis(cyclopentadienyl)titanium  
8      ethyl bromide, bis(cyclopentadienyl)titanium ethyl iodide,  
9      bis(cyclopentadienyl)titanium phenyl bromide,  
10     bis(cyclopentadienyl)titanium phenyl iodide,  
11     bis(cyclopentadienyl)zirconium methyl bromide,  
12     bis(cyclopentadienyl)zirconium methyl iodide,  
13     bis(cyclopentadienyl)zirconium ethyl bromide,  
14     bis(cyclopentadienyl)zirconium ethyl iodide,  
15     bis(cyclopentadienyl)zirconium phenyl bromide,  
16     bis(cyclopentadienyl)zirconium phenyl iodide; the trialkyl  
17     metallocenes such as cyclopentadienyltitanium trimethyl,  
18     cyclopentadienyl zirconium triphenyl, and cyclopentadienyl zirconium  
19     trineopentyl, cyclopentadienylzirconium trimethyl,  
20     cyclopentadienylhafnium triphenyl, cyclopentadienylhafnium  
21     trineopentyl, and cyclopentadienylhafnium trimethyl.

22       Other metallocenes which may be usefully employed to prepare  
23       the polymers of the invention include the monocyclopentadienyls  
24       titanocenes such as, pentamethylcyclopentadienyl titanium  
25       trichloride, pentaethylcyclopentadienyl titanium trichloride;  
26       bis(pentamethylcyclopentadienyl) titanium diphenyl, the carbene  
27       represented by the formula bis(cyclopentadienyl)titanium=CH<sub>2</sub> and  
28       derivatives of this reagent such as

29       bis(cyclopentadienyl)Ti=CH<sub>2</sub>.Al(CH<sub>3</sub>)<sub>3</sub>, (Cp<sub>2</sub>TiCH<sub>2</sub>)<sub>2</sub>,  
30       Cp<sub>2</sub>TiCH<sub>2</sub>CH(CH<sub>3</sub>)CH<sub>2</sub>, Cp<sub>2</sub>Ti-CHCH<sub>2</sub>CH<sub>2</sub>; substituted  
31       bis(cyclopentadienyl)titanium (IV) compounds such as:  
32       bis(indenyl)titanium diphenyl or dichloride,  
33       bis(methylcyclopentadienyl)titanium diphenyl or dihalides; dialkyl,  
34       trialkyl, tetra-alkyl and penta-alkyl cyclopentadienyl titanium  
35       compounds such as bis(1,2-dimethylcyclopentadienyl)titanium diphenyl  
36       or dichloride, bis(1,2-diethylcyclopentadienyl)titanium diphenyl or  
37       dichloride and other dihalide complexes; silicon, phosphine, amine or

1 carbon bridged cyclopentadiene complexes, such as dimethyl  
2 silyldicyclopentadienyl titanium diphenyl or dichloride, methyl  
3 phosphine dicyclopentadienyl titanium diphenyl or dichloride,  
4 methylenedicyclopentadienyl titanium diphenyl or dichloride and other  
5 dihalide complexes and the like.

6 Additional zirconocene catalysts useful according to the  
7 present invention include bis(cyclopentadienyl)zirconium dimethyl,  
8 bis(cyclopentadienyl)zirconium dichloride,  
9 bis(cyclopentadienyl)zirconium methylchloride,  
10 pentamethylcyclopentadienyl zirconium trichloride,  
11 pentaethylcyclopentadienyl zirconium trichloride,  
12 bis(pentamethylcyclopentadienyl)zirconium diphenyl, the alkyl  
13 substituted cyclopentadienes, such as bis(ethyl  
14 cyclopentadienyl)zirconium dimethyl,  
15 bis( $\beta$ -phenylpropylcyclopentadienyl)zirconium dimethyl,  
16 bis(methylcyclopentadienyl)zirconium dimethyl,  
17 bis(n-butyl-cyclopentadienyl)zirconium dimethyl,  
18 bis(cyclohexylmethylcyclopentadienyl)zirconium dimethyl,  
19 bis(n-octyl-cyclopentadienyl)zirconium dimethyl, and haloalkyl and  
20 dihalide complexes of the above; di-alkyl, trialkyl, tetra-alkyl, and  
21 pent-alkyl cyclopentadienes, such as  
22 bis(pentamethylcyclopentadienyl) zirconium di-methyl,  
23 bis(1,2-dimethylcyclopentadienyl)zirconium dimethyl and dihalide  
24 complexes of the above; silicone, phosphorus, and carbon bridged  
25 cyclopentadiene complexes such as dimethylsilyldicyclopentadienyl  
26 zirconium dimethyl or dihalide, and methylene dicyclopentadienyl  
27 zirconium dimethyl or dihalide, and methylene dicyclopentadienyl  
28 zirconium dimethyl or dihalide, carbenes represented by the formula  
29  $Cp_2Zr=CHP(C_6H_5)_2CH_3$ , and derivatives of these compounds  
30 such as  $Cp_2Zr\underset{CH_2}{CH}CH(CH_3)CH_2$ .

31 Bis(cyclopentadienyl)hafnium dichloride,  
32 bis(cyclopentadienyl)hafnium dimethyl, bis(cyclopentadienyl)vanadium  
33 dichloride and the like are illustrative of other metallocenes.

34 Various inorganic oxide supports may be used for supported  
35 catalyst systems to prepare polymers of the present invention. The  
36 polymerizations are generally carried in the temperature range of  
37 about 0-160°C or even higher but this range is not meant to be

1       exclusive for preparing the polymers of the invention which may be  
2       prepared by any technique resulting in the structure set forth.  
3       Atmospheric, sub-atmospheric, or super-atmospheric pressure  
4       conditions may exist for the polymerization using the metallocene  
5       catalyst described above. It is generally preferred to use catalyst  
6       compositions at a concentration so as to provide from about 1 ppm to  
7       about 5000 ppm, most preferably 10 ppm to 300 ppm, by weight of  
8       transition metal based on the weight of monomers in the  
9       polymerization of the ethylene polymers.

10       A slurry polymerization process may generally use  
11       sub-atmospheric or super-atmospheric pressures and temperatures in  
12       the range of 40-110°C. In a slurry polymerization, a suspension of  
13       solid, particulate polymer is formed in a liquid polymerization  
14       medium to which ethylene and comonomers and often hydrogen along with  
15       catalyst are added. The liquid employed in the polymerization medium  
16       can be alkane or cycloalkane, or an aromatic hydrocarbon such as  
17       toluene, ethylbenzene or xylene. The medium employed should be  
18       liquid under the conditions of polymerization and relatively inert.  
19       Preferably, hexane or toluene is employed.

20       In a modification, polymers of the present invention may be  
21       formed by gas-phase polymerization. A gas-phase process utilizes  
22       super-atmospheric pressure and temperatures in the range of about  
23       50°-120°C. Gas-phase polymerization can be performed in a stirred or  
24       fluidized bed of catalyst and product particles in a pressure vessel  
25       adapted to permit the separation of product particles from unreacted  
26       gases. Thermostated ethylene, comonomer (including diene), hydrogen  
27       and an inert diluent gas such as nitrogen can be introduced or  
28       recirculated so as to maintain the particles at a temperature of  
29       50°-120°C. Triethylaluminum may be added as needed as a scavenger of  
30       water, oxygen, and other adventitious impurities. Polymer product  
31       can be withdrawn continuously or semi-continuing at a rate such as to  
32       maintain a constant product inventory in the reactor. After  
33       polymerization and deactivation of the catalyst, the product polymer  
34       can be recovered by any suitable means. In commercial practice, the  
35       polymer product can be recovered directly from the gas phase reactor,  
36       freed of residual monomer with a nitrogen purge, and used without  
37       further deactivation or catalyst removal. The polymer obtained can

1 be extruded into water and cut into pellets or other suitable  
2 comminuted shapes. Pigments, antioxidants and other additives, as is  
3 known in the art, may be added to the polymer.

4 The molecular weight of polymer product obtained in  
5 accordance with this invention can vary over a wide range, as low as  
6 500 up to 2,000,000 or higher and preferably 1,000 to about 500,000.

7 For the production of polymer product having a narrow  
8 molecular weight distribution, it is preferable to deposit only one  
9 metallocene on to the inert porous support material and employ said  
10 support metallocene together with the alumoxane as the polymerization  
11 catalyst.

12 It is highly desirable to have for many applications, such  
13 as extrusion and molding processes, polyethylenes which have a broad  
14 molecular weight distribution of the unimodal and/or the multimodal  
15 type. Such polyethylenes evidence excellent processability, i.e.  
16 they can be processed at a faster throughput rate with lower energy  
17 requirements and at the same time such polymers would evidence  
18 reduced melt flow perturbations. Such polyethylenes can be obtained  
19 by providing a catalyst component comprising at least two different  
20 metallocenes, each having different propagation and termination rate  
21 constants for ethylene polymerizations. Such rate constants are  
22 readily determined by one of ordinary skill in the art.

23 The molar ratio of the metallocenes, such as, for example,  
24 of a zirconocene to a titanocene in such catalysts, can vary over a  
25 wide range, and in accordance with this invention, the only  
26 limitation on the molar ratios is the breadth of the Mw distribution  
27 or the degree of bimodality desired in the product polymer.

28 Desirably, the metallocene to metallocene molar ratio will be about  
29 1:100 to about 100:1, and preferably 1:10 to about 10:1.

### 30 Examples

31 In the Examples following the alumoxane employed was  
32 prepared by adding 76.5 grams ferrous sulfate heptahydrate in 4  
33 equally spaced increments over a 2 hour period to a rapidly stirred 2  
34 liter round-bottom flask containing 1 liter of a 13.1 weight percent  
35 solution of trimethylaluminum (TMA) in toluene. The flask was  
36 maintained at 50°C and under a nitrogen atmosphere. Methane produced  
37 was continuously vented. Upon completion of the addition of ferrous

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1 sulfate heptahydrate, the flask was continuously stirred and  
2 maintained at a temperature of 50°C for 6 hours. The reaction  
3 mixture was cooled to room temperature and was allowed to settle.  
4 The clear solution containing the alumoxane was separated by  
5 decantation from the insoluble solids.

6 Molecular weights were determined on a Water's Associates  
7 Model No. 150C GPC (Gel Permeation Chromatography). The measurements  
8 were obtained by dissolving polymer samples in hot trichlorobenzene  
9 and filtered. The GPC runs are performed at 145°C in  
10 trichlorobenzene at 1.0 ml/min flow using styragel columns from  
11 Perkin Elmer, Inc. 0.1 percent solutions (300 microliters of  
12 trichlorobenzene solution) were injected and the samples were run in  
13 duplicate. The integration parameters were obtained with a  
14 Hewlett-Packard Data Module.

15 Example 1

16 A 1-liter stainless steel pressure vessel, equipped with an  
17 incline blade stirrer, an external water jacket for temperature  
18 control, a septum inlet and vent line, and a regulated supply of dry  
19 ethylene and nitrogen, was dried and deoxygenated with a nitrogen  
20 flow. 500 cc of dry, degassed hexane directly into the pressure  
21 vessel 100 cc of purified 1,4-hexadiene was then charged into vessel  
22 with a gas tight syringe. 10.0 cc of 0.785 molar (in total aluminum)  
23 methyl alumoxane was injected into the vessel by a gas tight syringe  
24 through the septum inlet and the mixture was stirred at 1,200 rpm's  
25 and 60°C for 5 minutes at 0 psig of nitrogen.

26 Bis(n-butyl-cyclopentadienyl)zirconium dichloride (1 mg) dissolved in  
27 1.0 ml of dry, distilled toluene was injected through the septum  
28 inlet into the vessel. After 1 minute, ethylene at 60 psig was  
29 admitted and while the reaction vessel was maintained at 60°C. The  
30 ethylene was passed into the vessel for 10 minutes at which time the  
31 reaction was stopped by rapidly venting and cooling. 24.5 gms of  
32 ethylene/hexadiene copolymer was recovered after evaporation of the  
33 liquid component under nitrogen.

34 Example 1A

35 A 1-liter stainless steel pressure vessel, equipped with an  
36 incline blade stirrer, an external water jacket for temperature  
37 control, a septum inlet and vent line, and a regulated supply of dry

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1 ethylene and nitrogen, was dried and deoxygenated with a nitrogen  
2 flow. 500 cc of dry, degassed toluene and 100 cc of dry, degassed  
3 1,4-hexadiene was injected directly into the pressure vessel. 10.0  
4 cc of 0.785 molar (in total aluminum) methyl alumoxane solution in  
5 toluene was injected into the vessel by a gas tight syringe through  
6 the septum inlet and the mixture was stirred at 1,200 rpm's and 70°C  
7 for 5 minutes at 0 psig of nitrogen. Bis(n-butylcyclopenta-  
8 dienyl)zirconium dichloride (0.10 mg) dissolved in 1.0 ml of dry.  
9 distilled toluene was injected through the septum inlet into the  
10 vessel. After 1 minute, ethylene at 50 psig was admitted and while  
11 the reaction vessel was maintained at 70°C. The ethylene was passed  
12 into the vessel for 30 minutes at which time the reaction was stopped  
13 by rapidly venting and cooling. 9.6 gms of ethylene/1,4-hexadiene  
14 copolymer was recovered after evaporation of the liquid components  
15 under nitrogen.

16 Example 2

17 A 1-liter stainless steel pressure vessel, equipped with an  
18 incline blade stirrer, an external water jacket for temperature  
19 control, a septum inlet and vent line, and a regulated supply of dry  
20 ethylene and nitrogen, was dried and deoxygenated with a nitrogen  
21 flow. 500 cc of dry, degassed toluene, 50 cc of purified 1-hexene,  
22 and 50 cc of dry, degassed 1,4-hexadiene were injected directly into  
23 the pressure vessel. 10.0 cc of 0.785 molar (in total aluminum)  
24 methyl alumoxane solution in toluene was injected into the vessel by  
25 a gas tight syringe through the septum inlet and the mixture was  
26 stirred at 1,200 rpm's and 70°C for 5 minutes at 0 psig of nitrogen.  
27 Bis(n-butylcyclopentadienyl)zirconium dichloride (0.10 mg) dissolved  
28 in 0.10 ml of dry, distilled toluene was injected through the septum  
29 inlet into the vessel. After 1 minute, ethylene at 50 psig was  
30 admitted and while the reaction vessel was maintained at 70°C. The  
31 ethylene was passed into the vessel for 30 minutes at which time the  
32 reaction was stopped by rapidly venting and cooling. 5.1 gms of  
33 ethylene-1-hexene-1,4-hexadiene terpolymer was recovered after  
34 evaporation of the liquid components under nitrogen.

35 Example 2A

36 A 1-liter stainless steel pressure vessel, equipped with an  
37 incline blade stirrer, an external water jacket for temperature

- 25 -

1 control, a septum inlet and vent line, and a regulated supply of dry  
2 ethylene and nitrogen, was dried and deoxygenated with a nitrogen  
3 flow. 500 cc of dry, degassed hexane was added directly into the  
4 pressure flow. 100 cc of purified 4-vinylcyclohexene was charged  
5 into reaction vessel with a gas tight syringe. 10.0 cc of 0.785  
6 molar (in total aluminum) methyl alumoxane solution in toluene was  
7 injected into the vessel by a gas tight syringe through the septum  
8 inlet and the mixture was stirred at 1,200 rpm's and 70°C for 5  
9 minutes at 0 psig of nitrogen.  
10 Bis(n-butylcyclopentadienyl)zirconium dichloride (0.1 mg) dissolved  
11 in 1.0 ml of dry, distilled toluene was injected through the septum  
12 inlet into the vessel. After 1 minute, ethylene at 62 psig was  
13 admitted and while the reaction vessel was maintained at 60°C. The  
14 ethylene was passed into the vessel for 10 minutes at which time the  
15 reaction was stopped by rapidly venting and cooling. 25.6 gms of  
16 ethylene/4-vinyl cyclohexene copolymer polyethylene was recovered  
17 after evaporation of the liquid components under nitrogen.

18 Example 2B

19 A 1-liter stainless steel pressure vessel, equipped with an  
20 incline blade stirrer, an external water jacket for temperature  
21 control, a septum inlet and vent line, and a regulated supply of dry  
22 ethylene and nitrogen, was dried and deoxygenated with a nitrogen  
23 flow. 250 cc of dry, degassed toluene and 200 cc of purified  
24 4-vinylcyclohexene were injected directly into the pressure vessel.  
25 10.0 cc of 0.785 molar (in total aluminum) methyl alumoxane in  
26 toluene solution in toluene was injected into the vessel by a gas  
27 tight syringe through the septum inlet and the mixture was stirred at  
28 1,200 rpm's and 70°C for 5 minutes at 0 psig of nitrogen.  
29 Bis(n-butylcyclopentadienyl)zirconium dichloride (0.05 mg) dissolved  
30 in 0.050 ml of dry, distilled toluene was injected through the septum  
31 inlet into the vessel. After 1 minute, ethylene at 25 psig was  
32 admitted and while the reaction vessel was maintained at 70°C. The  
33 ethylene was passed into the vessel for 3 minutes at which time the  
34 reaction was stopped by rapidly venting and cooling. 7.3 gms of  
35 ethylene/4-vinylcyclohexene copolymer was recovered after evaporation  
36 of the liquid components under nitrogen.

37 The structural properties of the polymers of the Examples

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1      are reflected in the drawing figures and reported in the following  
2      Table. Data on commercial polymers are reported in the Table and  
3      displayed in the drawing figures for comparison.

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TABLE

RESIN	COMONOMER CONTENT (MOLE PERCENT)			$M_w \times 10^3$	$M_w/M_n$	COMPOSITION DISTRIBUTION	CLUSTER INDEX	MELTING BEHAVIOR DISTIRIBUTION PEAK H.P. (°C)
	TYPE	AMOUNT	TYPE					
EXAMPLE 1	1,4 HEXADIENE	2.3	-	-	70.5	1.75	-	NARROW
EXAMPLE 2	1,4 HEXADIENE	2.0	HEXENE-1	1.9	62.6	1.80	-1	NARROW
EXAMPLE 2A	4-VINYLCYCLO- HEXENE	0.4	-	-	95.3	1.87	NARROW	NARROW
3 - EXXON LL 3001	HEXENE-1	3.7	-	-	103.1	3.30	BROAD	12.1
4 - DOWLOX 2045	OCTENE-1	2.8	-	-	114.6	3.67	BROAD	9.5
5 - EXXON LL 1001	BUTENE-1	4.4	-	-	118.0	3.33	BROAD	11.7
6 - DUPONT SCLAIR 110	BUTENE-1	3.3	-	-	115.7	3.79	BROAD	5.0
7 - MITSUI 2020L	4-METHYL- PENTENE-1	3.5	-	-	92.3	2.6	BROAD	-
8 - LLDPE2	BUTENE-1	5.2	-	-	96.1	1.78	NARROW	9.6
								NARROW
								94.0

1. ANALYZED BUT ONLY LOW QUALITY SPECTRA OBTAINED - IMPRECISE RESULT.
2. LABORATORY PREPARATION WITH A SOLUBLE VANADIUM OXYCHLORIDE/ETHYL ALUMINUM SESQUICHLORIDE CATALYST.

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CLAIMS:

1. A copolymer composition from the polymerization of ethylene and at least one comonomer comprising an alpha, internal nonconjugated diene, said copolymer composition having incorporated in its structure at least 3 mole percent of units of said at least one comonomer, said copolymer composition having a cluster index of about 9 or less, and said copolymer composition having an alpha-olefin content less than 15 weight percent.
2. The copolymer composition of claim 1 having a molecular weight distribution of about 3.0 or less.
3. The copolymer composition of claim 2 having a molecular weight distribution of about 2.5 or less.
4. The copolymer composition of claim 3 having a molecular weight distribution of about 2.0.
5. The copolymer composition of claim 1 having a comonomer composition distribution wherein more than 55 weight percent of the copolymer is within 50 percent of the median comonomer content in mole percent, of said composition.
6. The copolymer composition of claim 5 having greater than 55 weight percent of the copolymers within 50 percent of the median comonomer content in mole percent.
7. A copolymer composition of claim 1 having a cluster index of about 7 or less.
8. A copolymer composition of claim 7 having a cluster index of about 5 or less.
9. The copolymer composition of claim 1 having a monomer composition comprising about 0.1-25 mole percent of said diene.

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10. The copolymer composition of claim 9 having a monomer composition comprising about 0.1-10 mole percent of said diene.

11. The copolymer composition of claim 9 having a monomer composition comprising 0.1-2 mole percent of said diene.

12. The copolymer composition of claim 1 wherein said copolymer is an interpolymer comprising ethylene, said diene, and at least one other polymerizable unsaturated comonomer.

13. The copolymer composition of claim 1 wherein said at least one other polymerizable unsaturated comonomer is an alpha-olefin.

14. The copolymer composition of claim 1 wherein said alpha-olefin is propylene, butene-1, or hexene-1.

15. The copolymer composition of claim 1 having a density of about 0.85-1.0.

16. The copolymer composition of claim 15 having a density of about 0.87-0.90.

17. The copolymer composition of claim 15 having a density of about 0.90-0.915.

18. The copolymer composition of claim 15 having a density of about 0.915-0.940.

19. The copolymer composition of claim 15 having a density of about 0.940-0.960.

20. The copolymer composition of claim 2 having a comonomer composition distribution wherein greater than 55 weight percent of the copolymer is within 50 percent of the median comonomer content in mole percent.

- 30 -

21. The copolymer composition of claim 1 wherein said copolymer has incorporated in its structure at least 5 mole percent of units of said at least one comonomer.

22. The copolymer composition of claim 1 wherein said at least 5 mole percent units of said at least one comonomer includes at least about 0.01 mole percent of units of said diene.

23. The copolymer composition of claim 22 containing at least about 0.1 mole percent units of said diene.

24. The copolymer composition of claim 23 containing at least about 1 mole percent units of said diene.

25. The copolymer composition of claim 24 containing at least about 3 mole percent units of said diene.

26. The copolymer composition of claim 25 containing at least 5 mole percent units of said diene.

27. The copolymer composition of claim 1 wherein said at least 3 mole percent units of said at least one comonomer consist essentially of said diene.

28. The copolymer composition of claim 1 wherein said diene is a linear diene.

29. The copolymer composition of claim 28 wherein said linear diene is a 1,4-hexadiene, substituted or unsubstituted.

30. The copolymer composition of claim 29 wherein said diene is 1,4-hexadiene.

31. The copolymer composition of claim 1 wherein said diene is a cyclic diene.

- 31 -

32. The copolymer composition of claim 1 wherein said cyclic diene is 4-vinylcyclohexene.

33. The copolymer composition of claim 27 wherein said diene is 1,4-hexadiene.

34. A copolymer composition of ethylene and an alpha, internal nonconjugated diene (diolefin) having a cluster index of about 9 or less, a molecular weight distribution of about 3.0 or less, a comonomer composition distribution of more than 55 weight percent of the copolymer within 50 percent of the median comonomer content in mole percent, said copolymer being formed by polymerization in the presence of a catalyst system comprising a metallocene of a metal of Group IVB, VB, and VIB of the Periodic Table and an alumoxane or a reaction product thereof, and said copolymer composition having an alpha-olefin content less than 15 weight percent.

35. The copolymer composition of claim 34 wherein the catalyst system metallocene is represented by

$(Cp)_m^M R_n R'_p$   
wherein Cp is a substituted or unsubstituted cyclopentadienyl ring; M is a Group IVB, VB, or VIB transition metal; R and R'; are independently selected halogen, hydrocarbyl group, or hydrocarboxyl group having 1-20 carbon atoms; m = 1-3, n = 0-3, p = 0-3, and the sum of m + n + p equals the oxidation state of M.

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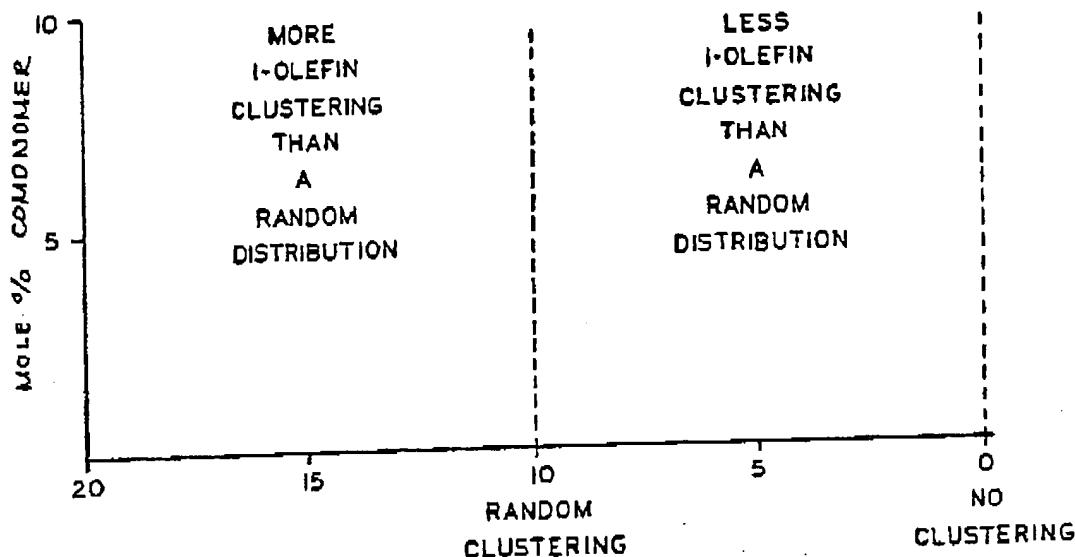


FIG.1

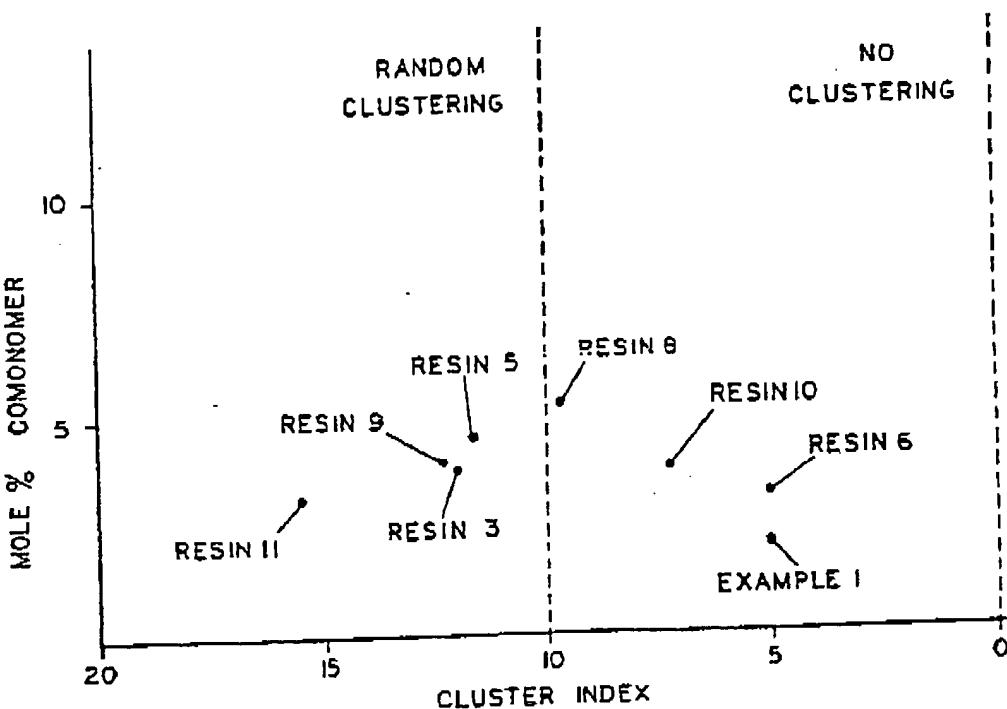


FIG.2

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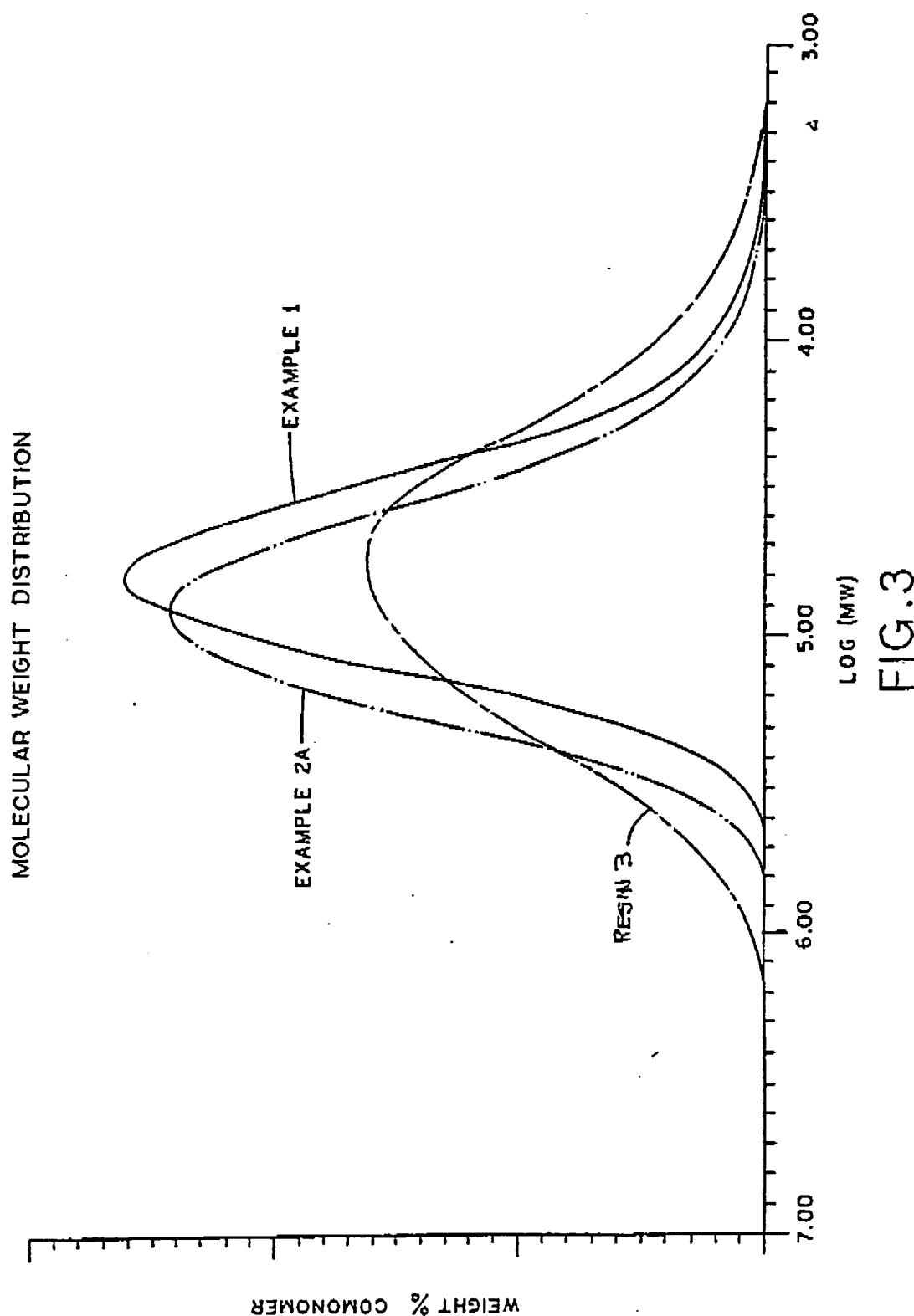


FIG. 3

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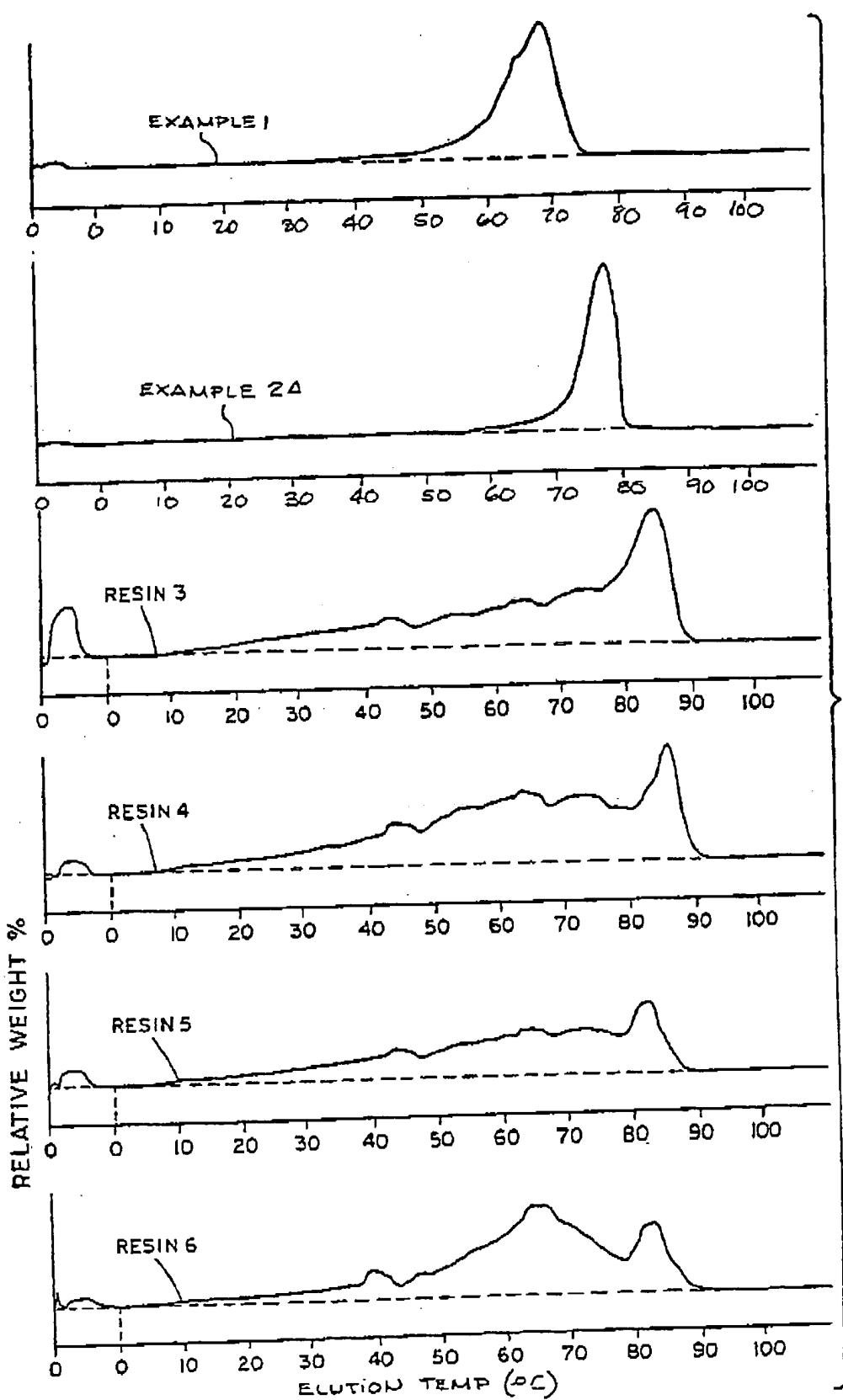


FIG.4

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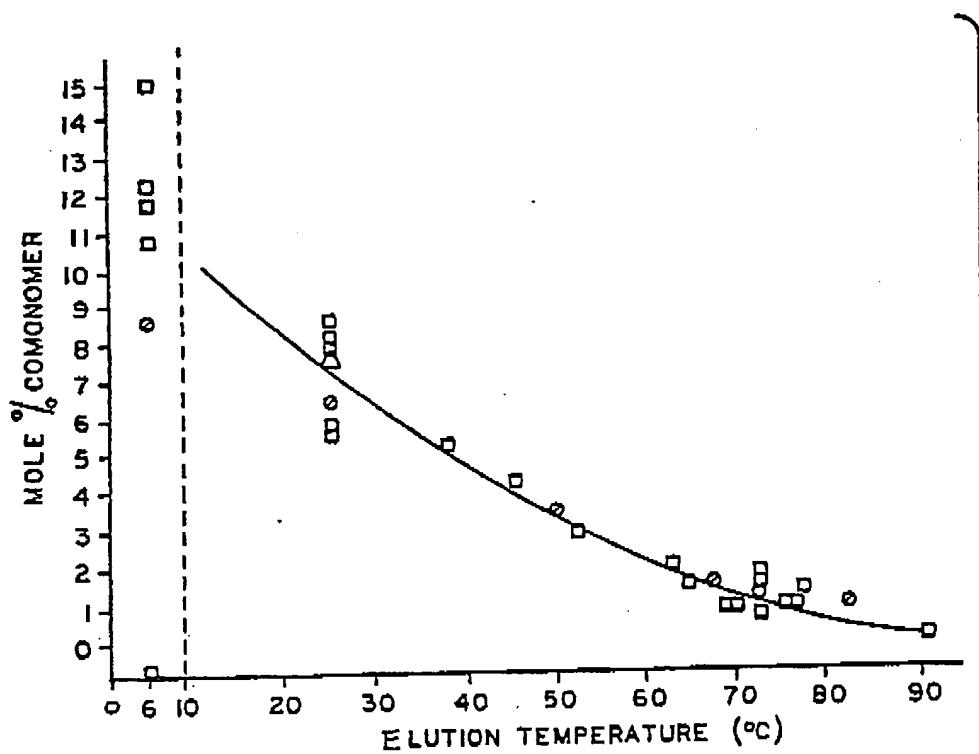


FIG.5

BUTENE COMONOMER	○
HEXENE COMONOMER	□
OCTENE COMONOMER	△

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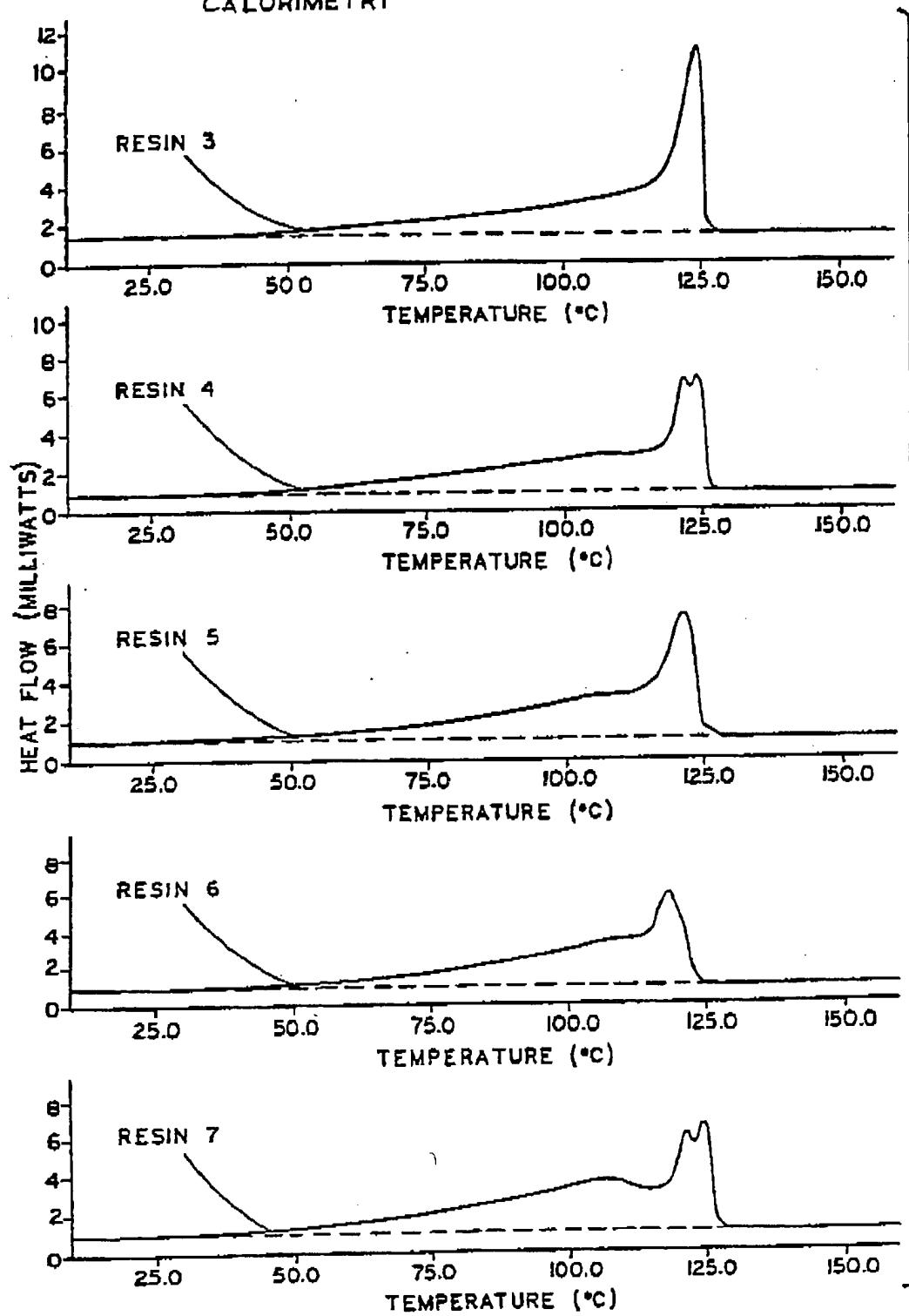
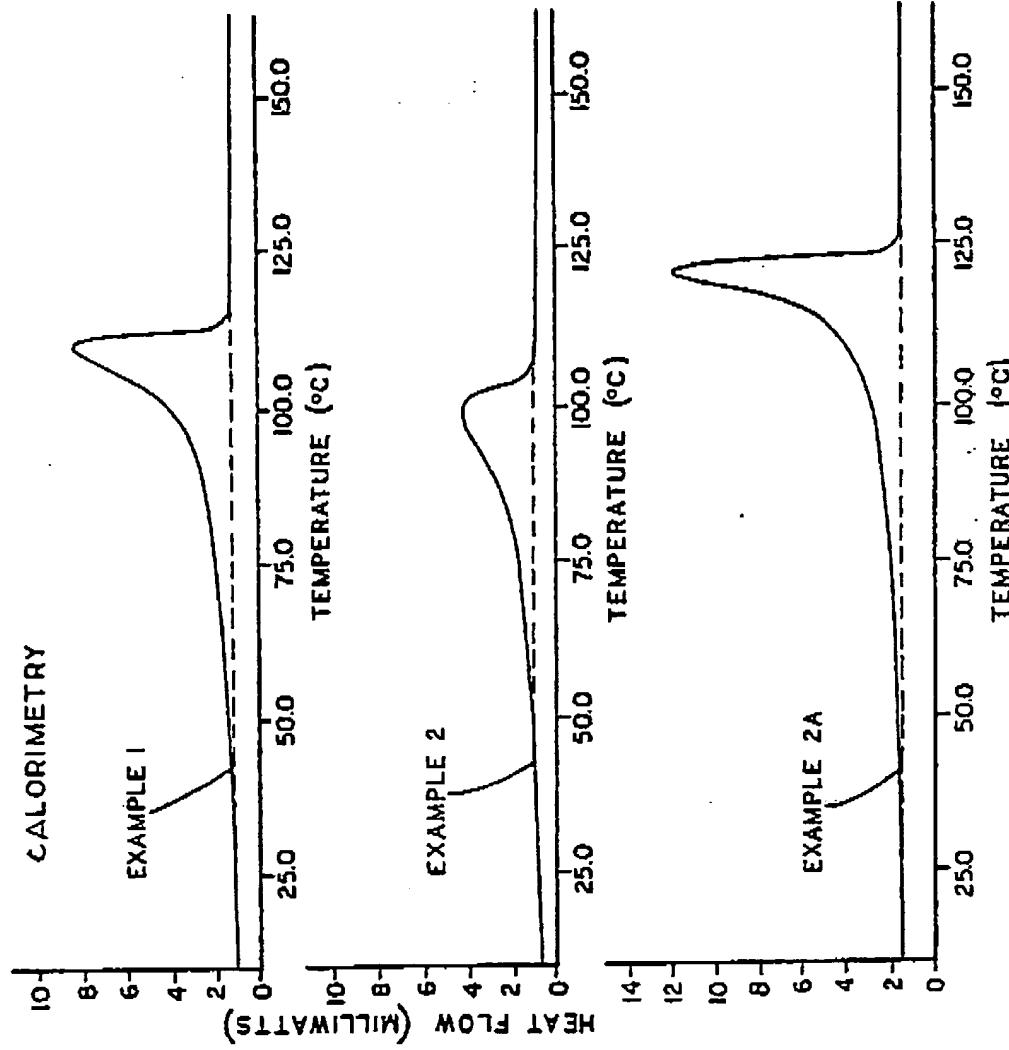
DIFFERENTIAL  
SCANNING  
CALORIMETRY

FIG. 6

FIG. 6A

DIFFERENTIAL  
SCANNING  
CALORIMETRY



# INTERNATIONAL SEARCH REPORT

International Application No. PCT/US 87/03299

## I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) \*

According to International Patent Classification (IPC) or to both National Classification and IPC

IPC<sup>4</sup>: C 08 F 210/18

## II. FIELDS SEARCHED

Minimum Documentation Searched ?

Classification System	Classification Symbols
IPC <sup>4</sup>	C 08 F

Documentation Searched other than Minimum Documentation  
to the Extent that such Documents are Included in the Fields Searched \*

## III. DOCUMENTS CONSIDERED TO BE RELEVANT\*

Category *	Citation of Document, <sup>11</sup> with indication, where appropriate, of the relevant passages <sup>12</sup>	Relevant to Claim No. <sup>13</sup>
A	EP, A, 0069951 (HOECHST) 19 January 1983 see claims 1-8 --	1
A	US, A, 4306041 (C. COZEWITH) 15 December 1981 see claims 1-19; column 3, lines 34-56 --	1
A	GB, A, 2099836 (MONTEPISON) 15 December 1982 see claims 1-7 --	1
P,X	EP, A, 0223394 (UNIROYAL CHEMICAL) 27 May 1987 see claims 1-17; page 3, line 42 - page 5, line 15 -----	1-35

- \* Special categories of cited documents: <sup>10</sup>
- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubt on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
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- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step
- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
- "A" document member of the same patent family

## IV. CERTIFICATION

Date of the Actual Completion of the International Search

15th April 1988

Date of Mailing of this International Search Report

02 JUN 1988

International Searching Authority

EUROPEAN PATENT OFFICE

Signature of Authorized Officer

P.C.G. VAN DER PUTTEN

**ANNEX TO THE INTERNATIONAL SEARCH REPORT  
ON INTERNATIONAL PATENT APPLICATION NO.**

US 8703299  
SA 20109

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on 11/05/88. The European Patent Office is in no way liable for those particulars which are merely given for the purpose of information.

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
EP-A- 0069951	19-01-83	DE-A-	3127133	27-01-83
		JP-A-	58019309	04-02-83
		CA-A-	1190998	23-07-85
		US-A-	4542199	17-09-85
US-A- 4306041	15-12-81	None		
GB-A- 2099836	15-12-82	BE-A-	893443	08-12-82
		FR-A-	2507189	10-12-82
		JP-A-	58002310	07-01-83
		NL-A-	8202251	03-01-83
		DE-A-	3221853	17-02-83
EP-A- 0223394	27-05-87	AU-A-	6397686	30-04-87
		JP-A-	62095304	01-05-87
		US-A-	4668834	26-05-87